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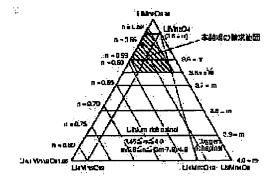
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(54) LITHIUM MANGANESE COMPOUND OXIDE, ITS PRODUCTION AND ITS USE

. (57) Abstract:

PROBLEM TO BE SOLVED: To produce a lithium manganese compound oxide for the positive electrode of a lithium cell.

SOLUTION: The lithium manganese compound oxide is a compd. consisting of Li, Mn and O, represented by the formula Li1+xMn2-y04 (where -0.01 < x < 0.15and 0 < v < 0.15) and having a cubic spinel structure. The atomic ratio of Li to Mn is 0.52-0.59 and the average oxidation number of Mn is 3.45-3.65. The multiple oxide has 0.821-0.824 nm lattice constant, 60-180 nm crystallite diameter and 1.0-3.7 m2/g BET specific surface area, contains at least $\geq 3\%$ primary particles having $\geq 1~\mu$ m particle diameter and has 1.0-15.0 μ m median diameter on the particle size distribution curve measured by a laser diffraction scattering method, an aggregation index of 5-20 and ≥55% press molding density.



LIVIngCade, LittleMausOctus - Lister + Co 第三角ダイアグラム。 による非化学風味組成のスピネル化合物の表示

LEGAL STATUS

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